

INSTRUCTION MANUAL FOR CHIRALPAK® IA, CHIRALPAK® IB, CHIRALPAK® IC, CHIRALPAK® ID, CHIRALPAK® IE, CHIRALPAK® IF and CHIRALPAK IG

<Reverse Phase>

Please read this instruction sheet completely before using these columns

These columns can also be used in normal phase mode.
 Please refer to the corresponding instruction sheet for details.

Switching Between RP and NP Mode

To switch from reversed phase mode to normal phase mode, and vice versa, column should be carefully flushed with miscible solvent.
 It is highly recommended to apply the **regeneration procedure** described in the instruction sheet for normal phase mode. Before applying this protocol, any traces of salts should be removed by flushing with a mobile phase that does not contain any salts / buffers.

Method Development / Reversed Phase

A - Mobile phases / For both UV and Mass detections

		ACIDIC (AMPHOTERIC) Compounds	NEUTRAL Compounds	BASIC Compounds ❶
CHIRALPAK® IA CHIRALPAK® ID CHIRALPAK® IE CHIRALPAK® IF CHIRALPAK® IG CHIRALPAK® IB CHIRALPAK® IC	Aqueous solution ❶	HCOOH aq. pH 2.0	Water	20 mM NH ₄ HCO ₃ aq. pH 9.0 adjusted with a basic additive ❶
	Organic modifier ❷	CH ₃ CN or MeOH or EtOH or IPA or THF		
	Typical starting conditions ❸	Aqueous solutions CH ₃ CN	60%	40% ❹

☞ NOTE 1: If you cannot achieve sufficient resolution, try the complementary aqueous solutions

B - Complementary aqueous and buffer solutions / For UV detection

		ACIDIC (AMPHOTERIC) Compounds	NEUTRAL Compounds	BASIC Compounds ❶
CHIRALPAK® IA CHIRALPAK® ID CHIRALPAK® IE CHIRALPAK® IF CHIRALPAK® IG CHIRALPAK® IB CHIRALPAK® IC	Aqueous solution ❶	50 mM Phosphate Buffer pH 2.0 OR H ₃ PO ₄ aq. pH 2.0 OR 100 mM KPF ₆ (or NaPF ₆) aq. pH 2.0 adjusted with H ₃ PO ₄	Water	20 mM Borate Buffer pH 9.0 OR 20 mM Phosphate Buffer pH 8.0 ❷ OR 100 mM KPF ₆ (or NaPF ₆) aq.

☞ NOTE 2: The concentration of all the buffering salt should be less than 500 mM.

- ❶ Refer to **section C** for preparation of aqueous solution and choice of basic additives.
- ❷ ☐ It is recommended to use CH₃CN to start the investigation

- ❑ The elution power of organic modifiers for these columns is in the descending order of $\text{CH}_3\text{CN} > \text{EtOH} > \text{MeOH}$: $50\% \text{CH}_3\text{CN} \approx 65\text{-}70\% \text{EtOH} \approx 75\text{-}80\% \text{MeOH}$.
 - ❑ The use of other organic solvents –**except THF**– has not been investigated and could be harmful to the columns.
 - ❑ The use of alcohols causes the back pressure to be significantly higher compared to CH_3CN due to their high viscosity in mixtures with water.
- ③
- ❑ Retention can be adjusted by changing the proportion of CH_3CN . Retention may be very sensitive to the amount of CH_3CN present into the mobile phase.
 - ❑ Lowering the column temperature may increase the retention time and the selectivity.
 - ❑ Increasing the column temperature and decreasing the flow rate may increase the resolution.
- ④
- ❑ To maximize column life the use of a guard cartridge is essential when basic conditions are employed.
 - ❑ The use of strong basic conditions ($> \text{pH } 9$) must be avoided, as they are known to damage the silica gel matrix.
 - ❑ When these columns are used at $\text{pH} > 7$, **the temperature should be maintained between 5°C and 25°C for maximum column life.**
- ⑤ High percentages of organic modifier in the mobile phase **may precipitate the buffering salt** from the solution, and lead to consequent clogging of the column (refer to the table below).

Water / Organic Modifier	Buffer solution / Organic Modifier
90 / 10 to 0 / 100	90 / 10 to 15 / 85

- ⑥ Do not use the phosphate buffer for $\text{pH} > 8$. When $\text{pH } 9$ is necessary, use the ammonium bicarbonate solution or borate buffer for maximum column life.

C – Buffer preparation – Examples

➤ Preparation of pH 2 Phosphate buffer:

Solution A: 50 mM potassium dihydrogenphosphate
3.40g KH_2PO_4 / FW 136.09, make up the volume to 500ml with HPLC grade water

Solution B: phosphoric acid (H_3PO_4 85% by weight)
Adjust the pH of solution A to a value of 2.0 using solution B.

➤ Preparation of pH 2 KPF_6 (NaPF_6) solution:

Solution A: 100 mM potassium (sodium) hexafluorophosphate
9.20g KPF_6 / FW 184.06 or 8.40g NaPF_6 / FW 167.95, make up the volume to 500 ml with HPLC grade water

Solution B: phosphoric acid (H_3PO_4 85% by weight)
Adjust the pH of solution A to a value of 2.0 using solution B.

➤ Preparation of pH 9 Ammonium bicarbonate solution:

Solution A: 20 mM ammonium bicarbonate
0.78g NH_4HCO_3 / FW 78.05, make up the volume to 500 ml with HPLC grade water

Solution B: Basic additive such as diethylamine (DEA), triethylamine (TEA), ammonia (NH_3) and so on.
** DEA tends to give better peak shape than other bases.*

Adjust the pH of solution A to a value of 9.0 using solution B.

➤ Preparation of pH 8 Phosphate buffer:

Solution A: 20 mM potassium hydrogenophosphate
1.74g of K_2HPO_4 / FW 174.18, make up the volume to 500 ml with HPLC grade water

Solution B: 20 mM potassium dihydrogenophosphate
1.36g KH_2PO_4 / FW 136.09, make up the volume to 500 ml with HPLC grade water.

Adjust the pH of solution A to a value of 8.0 using solution B.

➤ Preparation of pH 9 Borate buffer:

Solution A: 20 mM sodium tetraborate decahydrate
3.81g of $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$ / FW 381.37, make up the volume to 500 ml with HPLC grade water

Solution B: 20 mM boric acid
0.62g H_3BO_3 / FW 61.83, make up the volume to 500 ml with HPLC grade water

Adjust the pH of solution A to a value of 9.0 using solution B.

Column Care / Maintenance

- Any traces of salts should be removed before column storage and /or before switching to 100% organic solvent (use Water/CH₃CN 60:40 (v/v) for instance)

Refer main instruction for normal phase and column care/maintenance.

⇒ If you have any questions about the use of these columns, or encounter a problem, contact:

In the USA: questions@chiraltech.com or call 800-6-CHIRAL

In the EU: cte@chiral.fr or call +33 (0)3 88 79 52 00

In India: chiral@chiral.daicel.com or call +91-40-2338-3700

Locations:

North/Latin America

Chiral Technologies, Inc.
800 North Five Points Road
West Chester, PA 19380
800 6 CHIRAL
Tel: 610-594-2100
Fax: 610-594-2325
chiral@chiraltech.com
www.chiraltech.com

Europe

Chiral Technologies Europe
Parc d'Innovation
Bd Gonthier d'Andernach
67400 Illkirch Cedex, France
Tel: +33-388-795-200
Fax: +33-388-667-166
cte@chiral.fr
www.chiral.fr

India

Daicel Chiral Technologies (India) Pvt. Ltd.
Lab No. 4A, Phase III
IKP Knowledge Park
Genome Valley, Turkapally,
Shameerpet, Ranga Reddy Dist.
Hyderabad-500 078, Telangana
Tel: +91-40-2338-3700
Fax: +91-40-2348-0104
chiral@chiral.daicel.com

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